

# Accumulation of Contaminants in the Distribution Systems

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**ORD, NRMRL, WSWRD, TTEB**

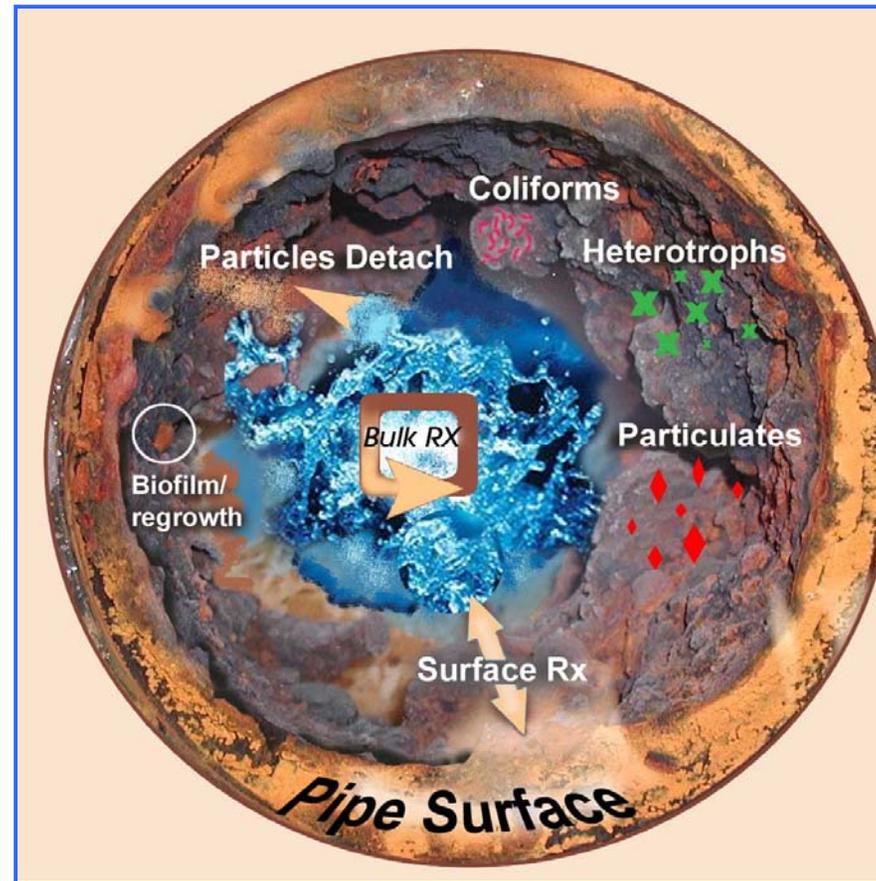
**Cincinnati, Ohio 45268**

2007 U.S. EPA, ORD/OGWDW, Workshop on Inorganic Contaminant Issue

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Millennium Hotel, Cincinnati, Ohio

# Corroding Pipes are Complicated Reactors that can Accumulate Trace Contaminants



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# General Nature of Metallic Pipe Surfaces

- Oxides, hydroxides, hydroxycarbonates, carbonates, hydroxysulfates, etc. from corrosion
- Similar compounds from deposition or post-precipitation (particularly Fe, Mn, Al), may include silicates
- Phosphates from corrosion control
- All may be mixed with NOM
- Biofilms present in some areas and some materials



# Practical Issues of Contaminant Accumulation

- What contaminants are involved (health risk?)
- What is the “equilibrium” mass of deposit?
- Where are the contaminants located
  - *Relative to consumer ingestion?*
  - *Relative to regulatory monitoring locations?*
  - *Relative to types of mains/pipes?*



# Accumulation of Contaminants in the DS



**Corrosion deposits, sediment, and other solids that collect in the DS can accumulate contaminants if in the water**



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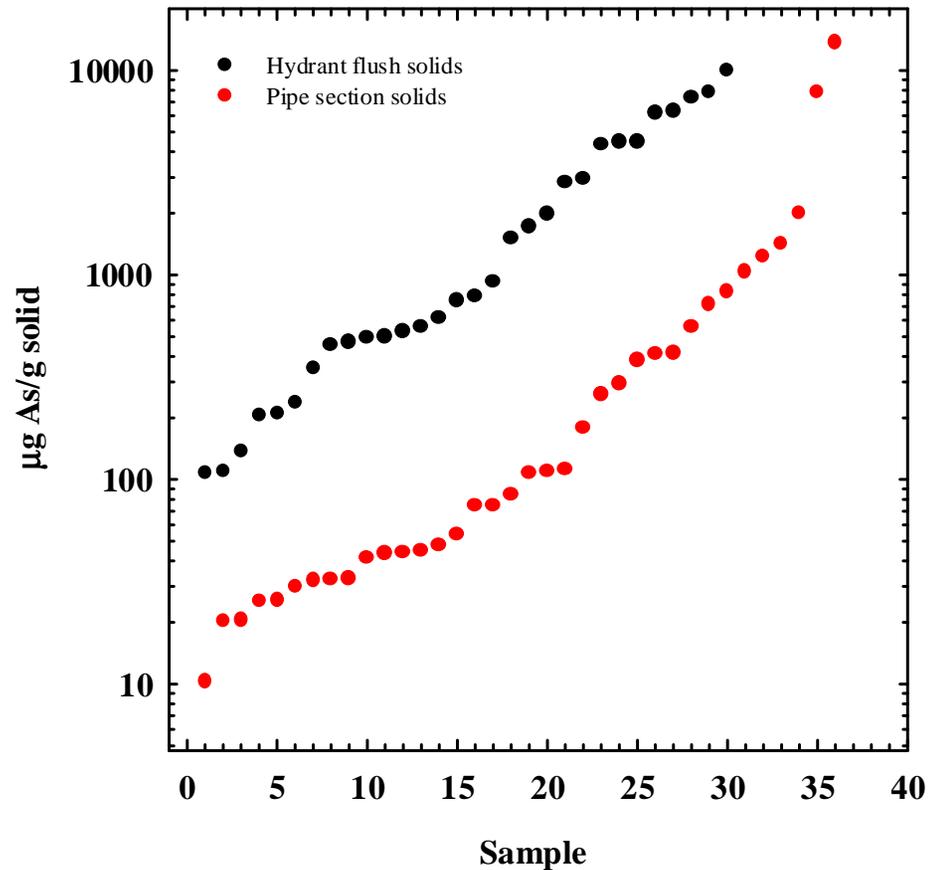
# Solids Analysis

- Acid digestion/ICP-MS (Battelle)
  - *Ca, Mn, Fe, Mg, P, Si, As, others*
  - *Units*
- XRD
  - *Mineral phases*
- Electron microprobe-WDS (Battelle)
  - *Quantitative elemental mapping*
- SEM-Wavelength dispersive spectrometer- imaging and elemental mapping



# Arsenic Accumulation in the DS

## ICP-MS Analysis

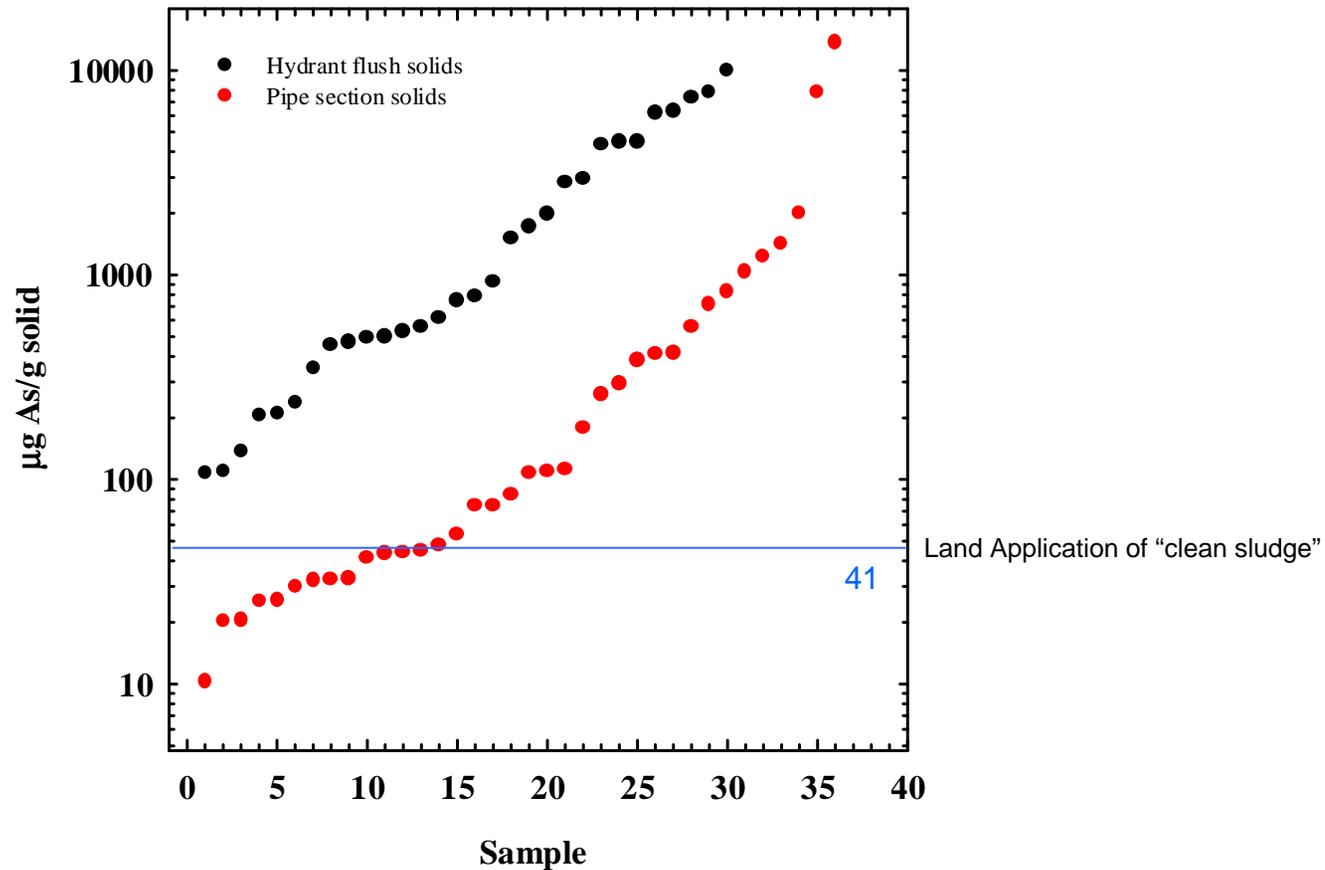


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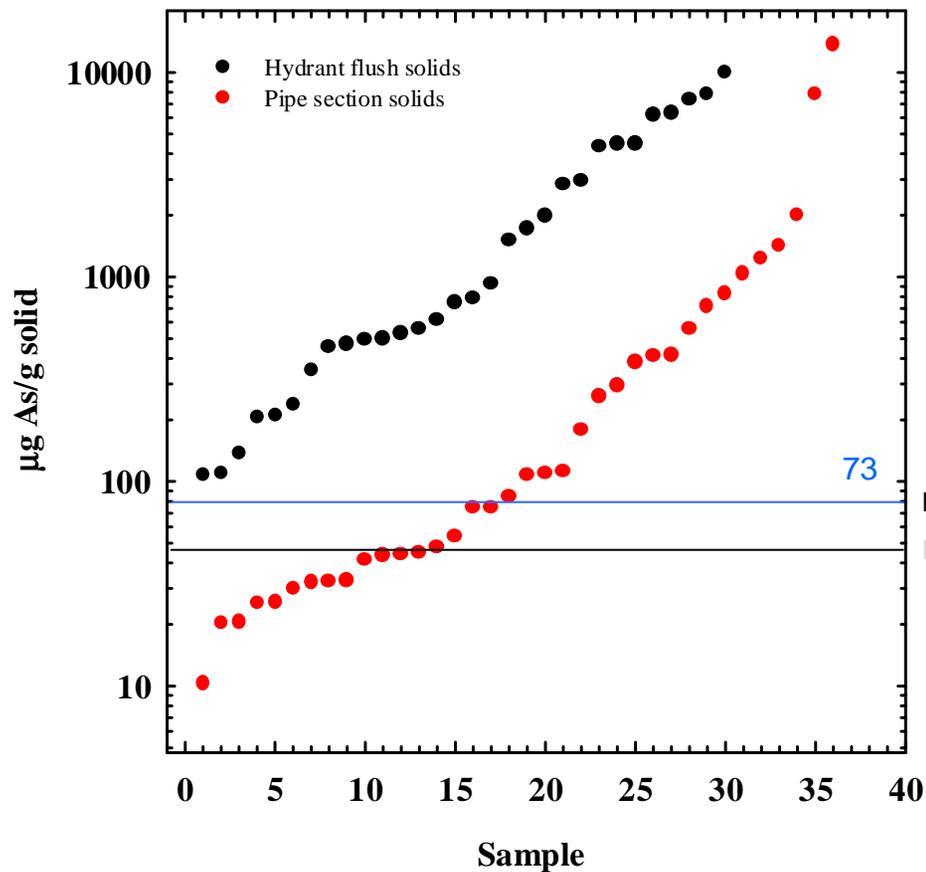
# Arsenic Accumulation in the DS

## ICP-MS Analysis



# Arsenic Accumulation in the DS

## ICP-MS Analysis



Land Disposal Limit, Dedicated Land Disposal sites

Land Application of "clean sludge"

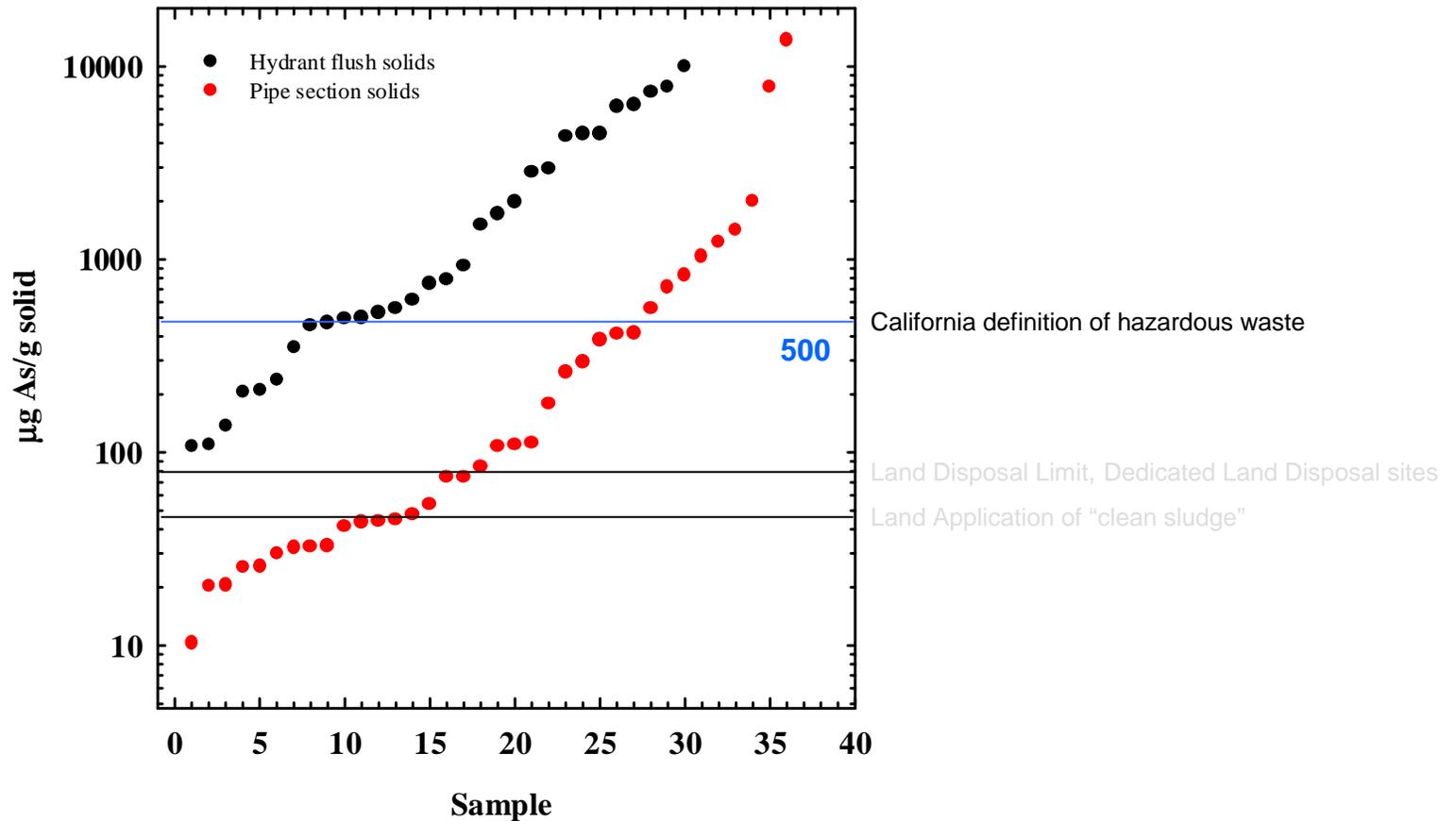


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# Arsenic Accumulation in the DS

## ICP-MS Analysis

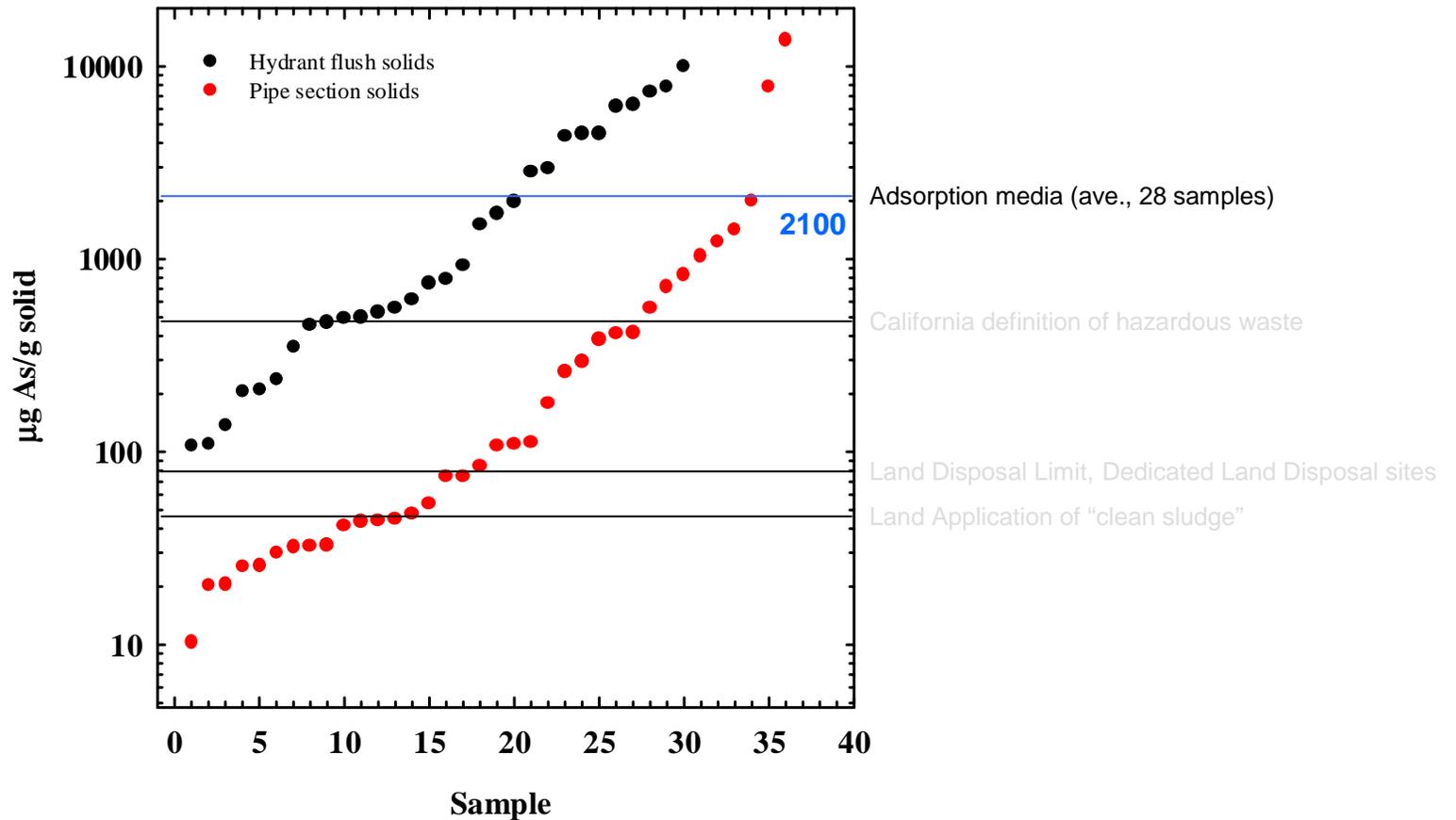


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# Arsenic Accumulation in the DS

## ICP-MS Analysis

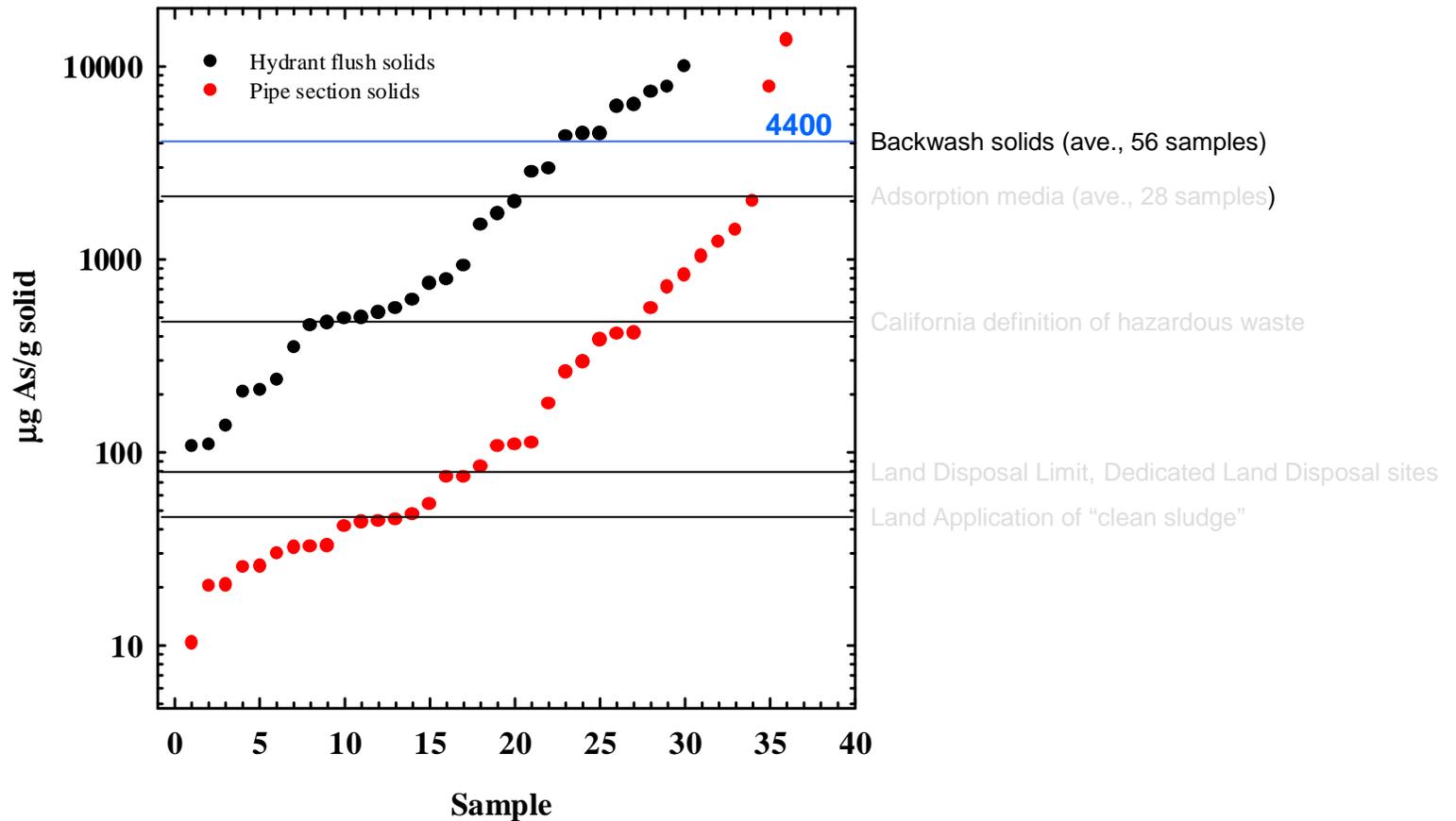


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# Arsenic Accumulation in the DS

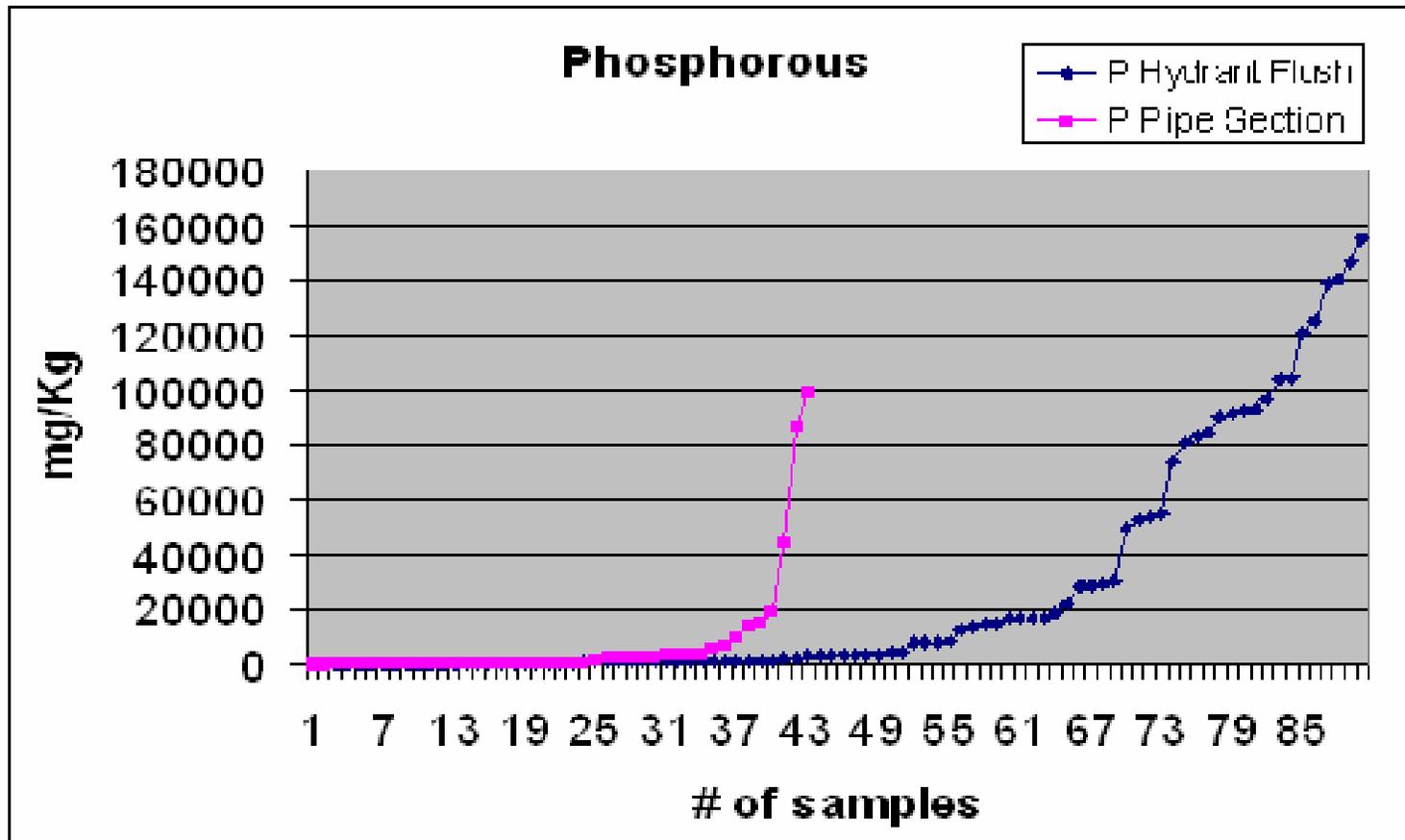
## ICP-MS Analysis



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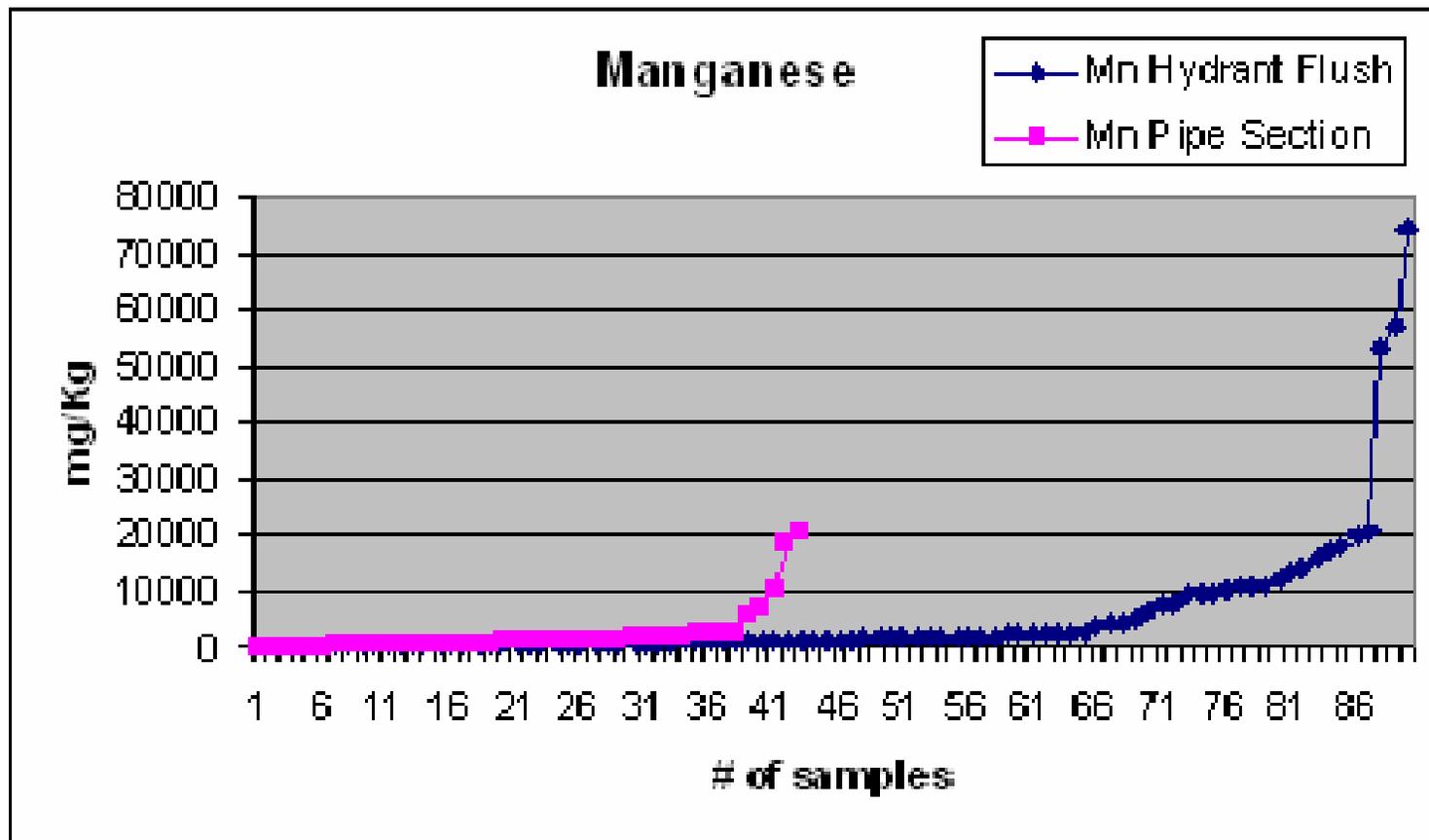
# Phosphorus Accumulation



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# Manganese Accumulation



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# Practical Issues of Contaminant Accumulation

- How easily is it destabilized?
  - *How is the contaminant bound?*
    - Solid mineral phase?
    - Sorbed?
  - *Phosphates: Do they*
    - “Seal” the surfaces?
    - Dissolve or displace the surface compounds and layers?
    - Mobilize sediment particles?
    - Any or all of the above?
  - *Hydraulic factors: pressure or flow changes*



# Particle Mobilization

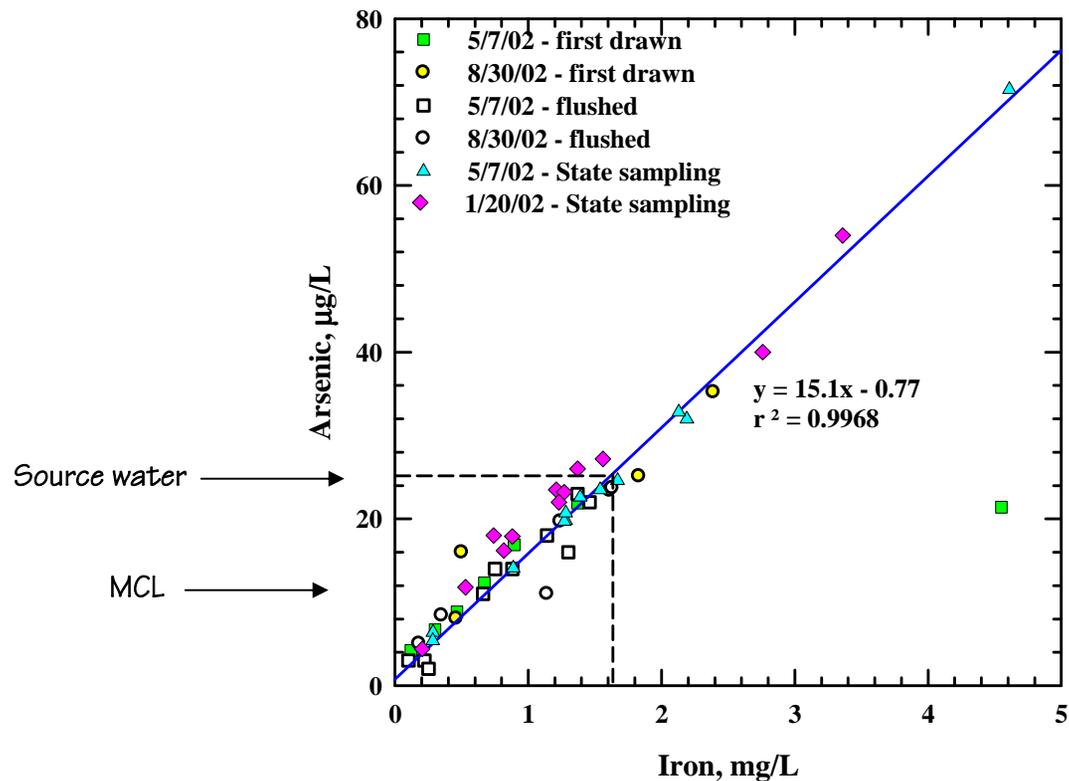
## Case Study 1: Particulate Release of Arsenic in Distribution Systems

- Colored water events led to sampling and the finding that As levels ( $>100 \mu\text{g/L}$ )
- Also high iron levels ( $>15 \text{ mg/L}$ )
- Lawsuit and media attention
- $73 \text{ mg Ca/L}$ ,  $32 \text{ mg Mg/L}$ ,  
 $17 \text{ mg SiO}_2/\text{L}$ ,  
pH mid 7's
- $24 \mu\text{g As/L}$ ,  $1.6 \text{ mg Fe/L}$
- Chlorination



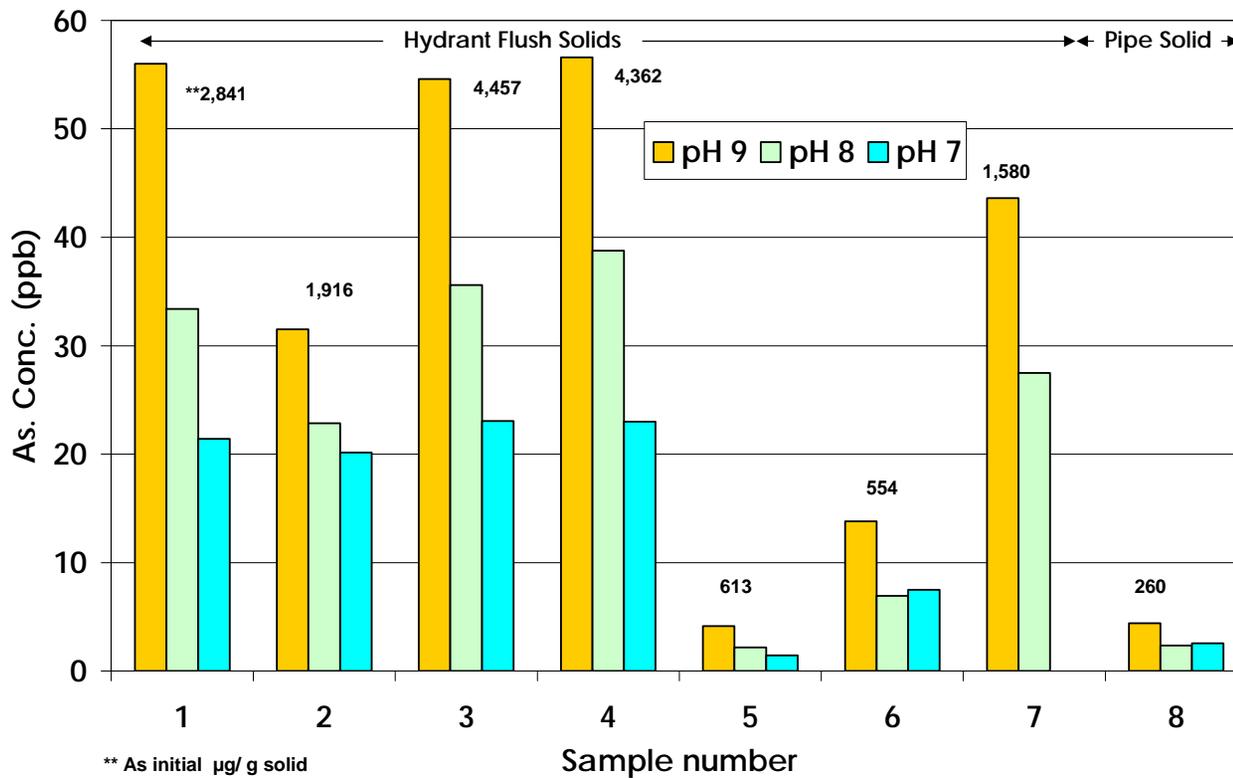
# Particle Mobilization

## Case Study 1: Relationship Between Arsenic and Iron in Distribution System Samples



# Desorption

## Research: Desorption from Drinking Water Distribution System Solids

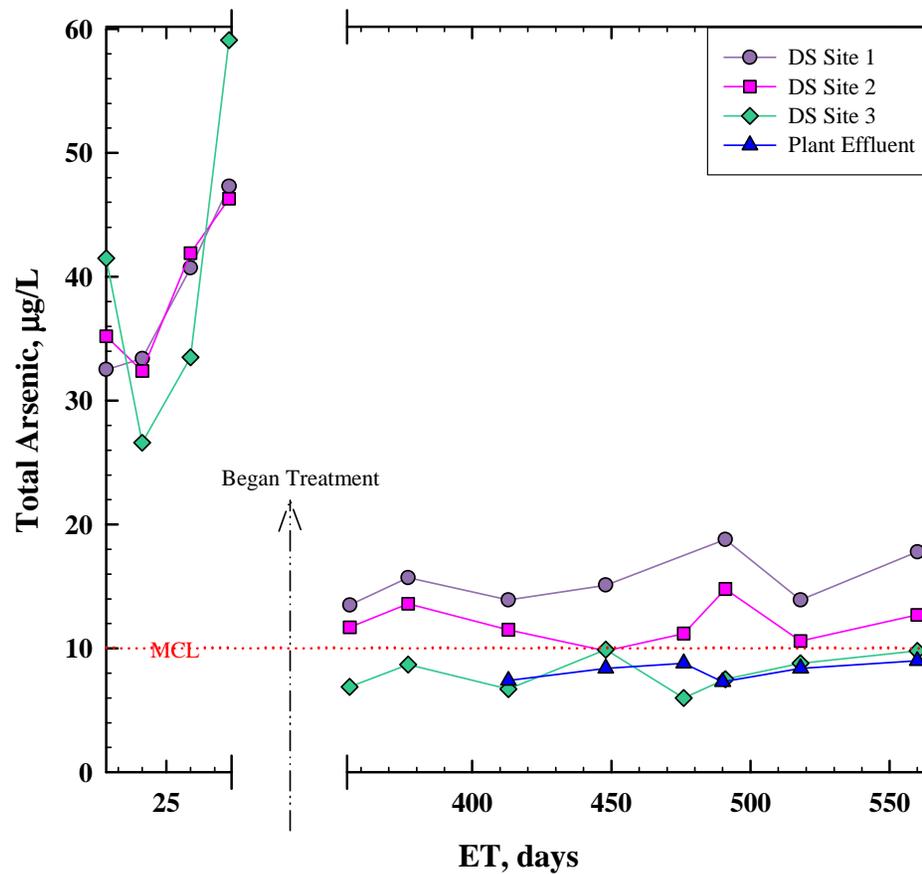


- Samples 1, 2,3,4 and 7 correspond to the same Utility
- The majority of these solids are hydrant flush material



# Desorption

## Lidgerwood, ND



**Initial soluble As concentration:**

**125 µg/L (reduced to approximately 35 µg/L prior to DS)**

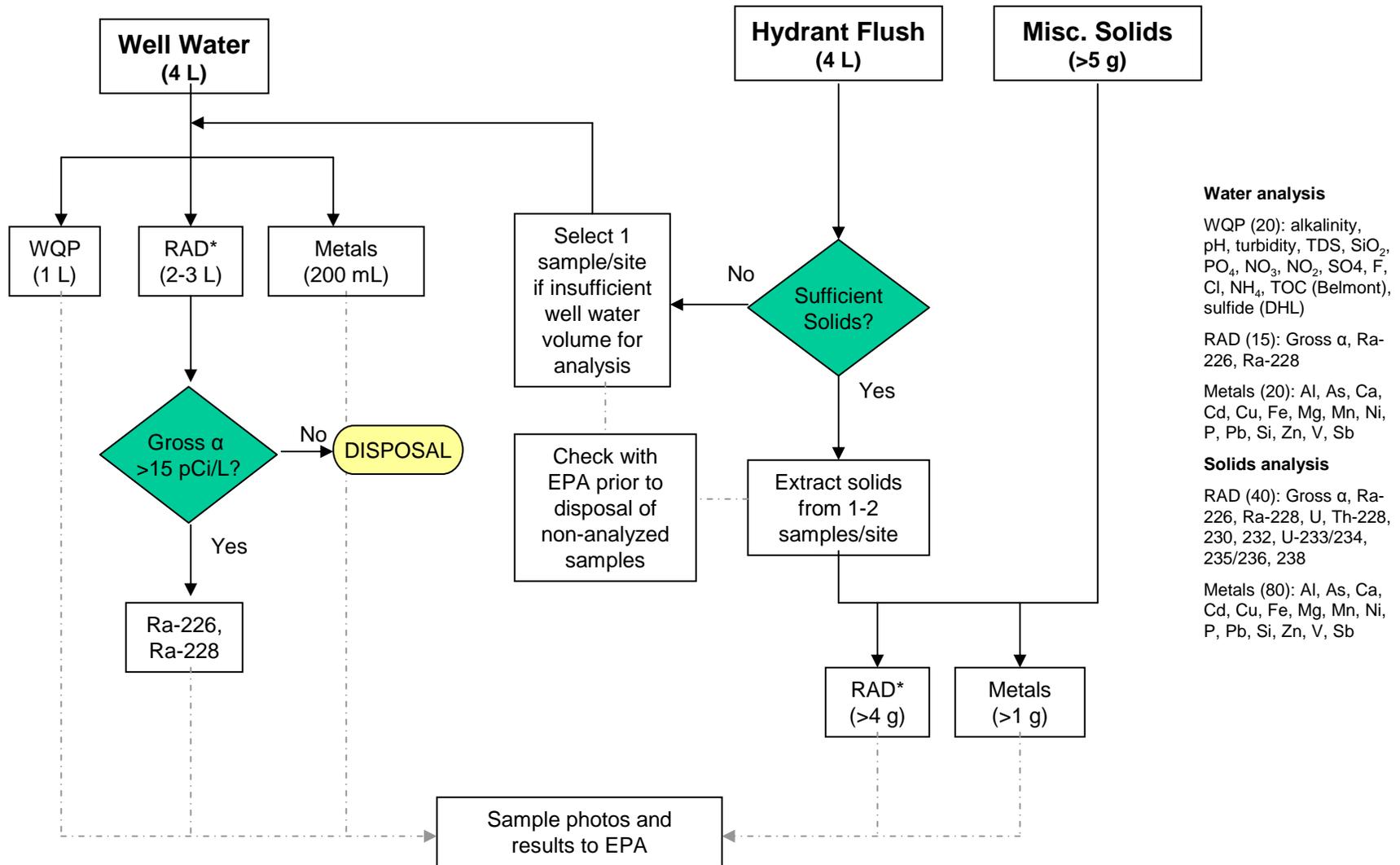


# Accumulation of Radium and Uranium in Distribution Systems

- Battelle Contractor
- Pipe sections and fire hydrant flush samples
- Midwest, Texas water systems
- Status: samples concentrated currently being analyzed



# Radium and Uranium Sample Flow Chart



## Water analysis

WQP (20): alkalinity, pH, turbidity, TDS, SiO<sub>2</sub>, PO<sub>4</sub>, NO<sub>3</sub>, NO<sub>2</sub>, SO<sub>4</sub>, F, Cl, NH<sub>4</sub>, TOC (Belmont), sulfide (DHL)

RAD (15): Gross α, Ra-226, Ra-228

Metals (20): Al, As, Ca, Cd, Cu, Fe, Mg, Mn, Ni, P, Pb, Si, Zn, V, Sb

## Solids analysis

RAD (40): Gross α, Ra-226, Ra-228, U, Th-228, 230, 232, U-233/234, 235/236, 238

Metals (80): Al, As, Ca, Cd, Cu, Fe, Mg, Mn, Ni, P, Pb, Si, Zn, V, Sb

WQP = water quality parameter analyses; RAD = radiological isotopes analyses; Metals = ICP/MS analyses; TBD = to be determined.

\* = first priority for analysis.

All samples will be screened by Radiation Safety Services for proper handling and storage guidelines.



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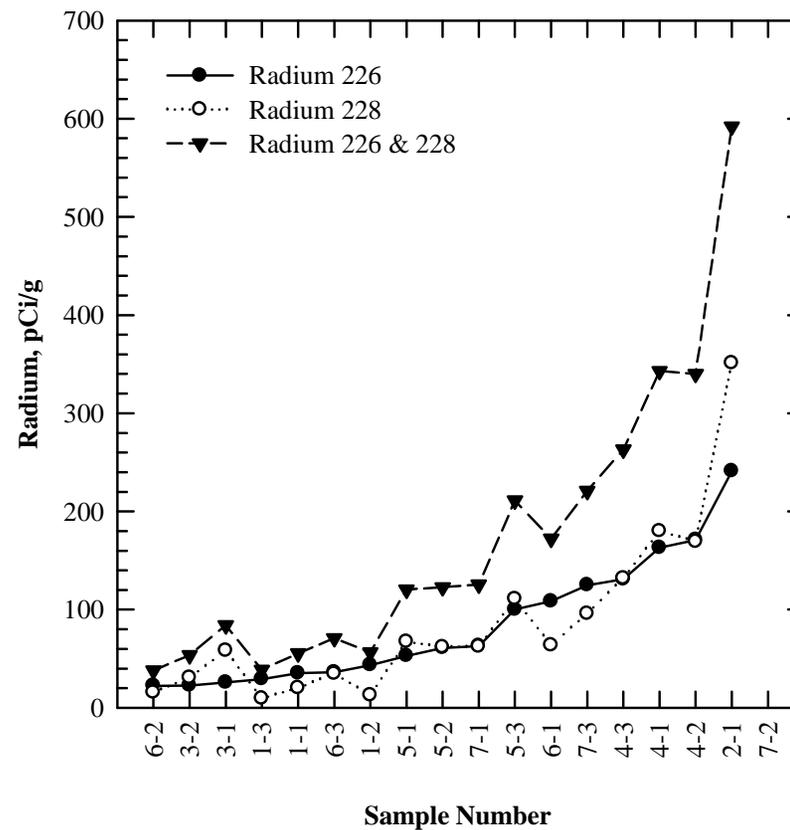
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# Accumulation of Contaminants in the DS

- Mg, Al, Si, P, Ca, V, Mn, Fe, Ni, Cu, Zn, As, Cd, Sb, Ba, Pb
- Gross  $\alpha$ ; gross  $\beta$ ; Ra-226 & 228; Th- 228, 230 & 232; U- 233/234, 235, 236 & 238
- 18 sample from 7 location in Midwest
  - 13 to 698 pCi/g gross  $\alpha$
  - 22 to 637 pCi/g gross  $\beta$
  - 23 to 241 pCi/g radium-226
  - 16 to 351 pCi/g radium-228



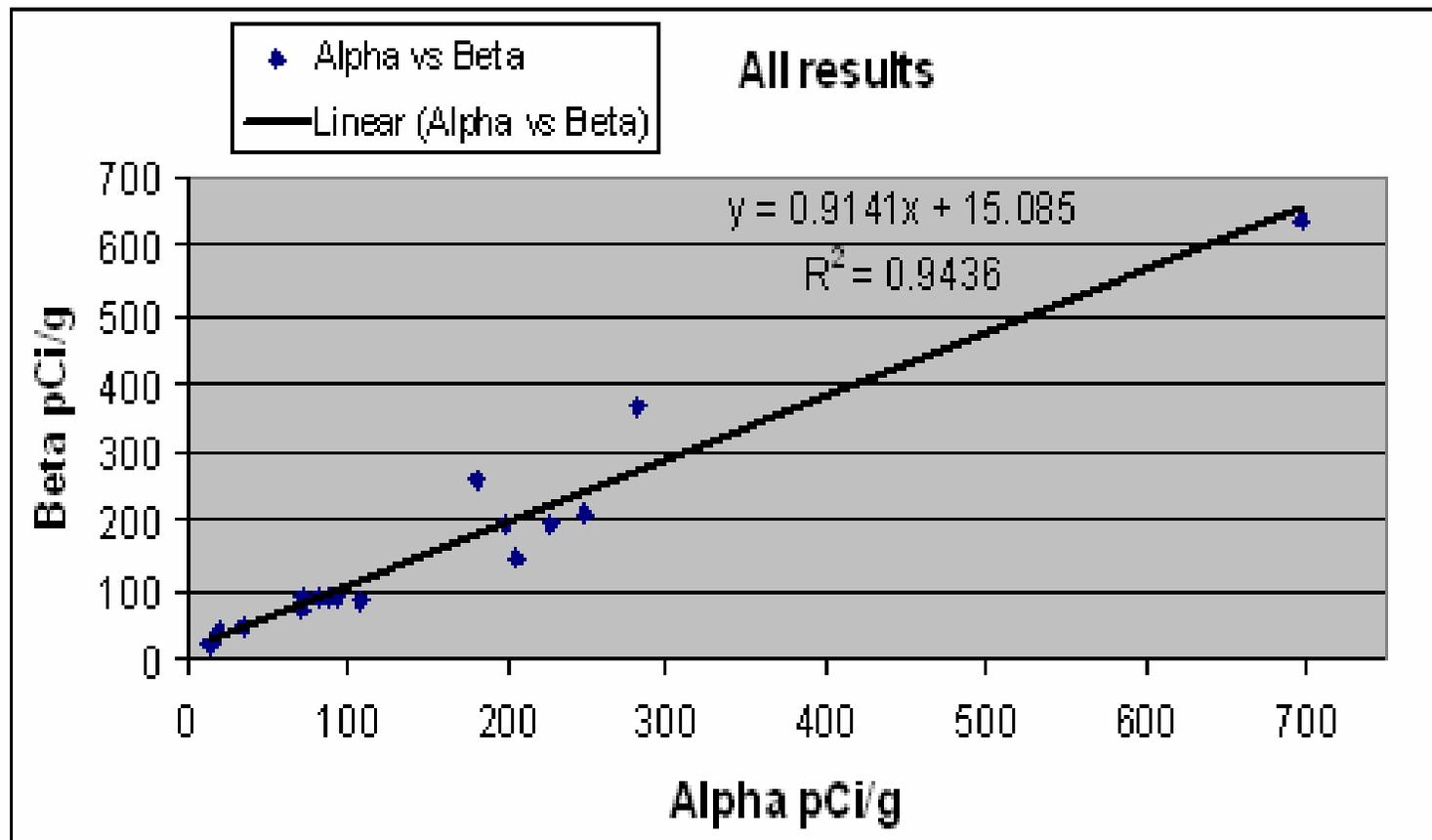
# Radium in Distribution System Solids



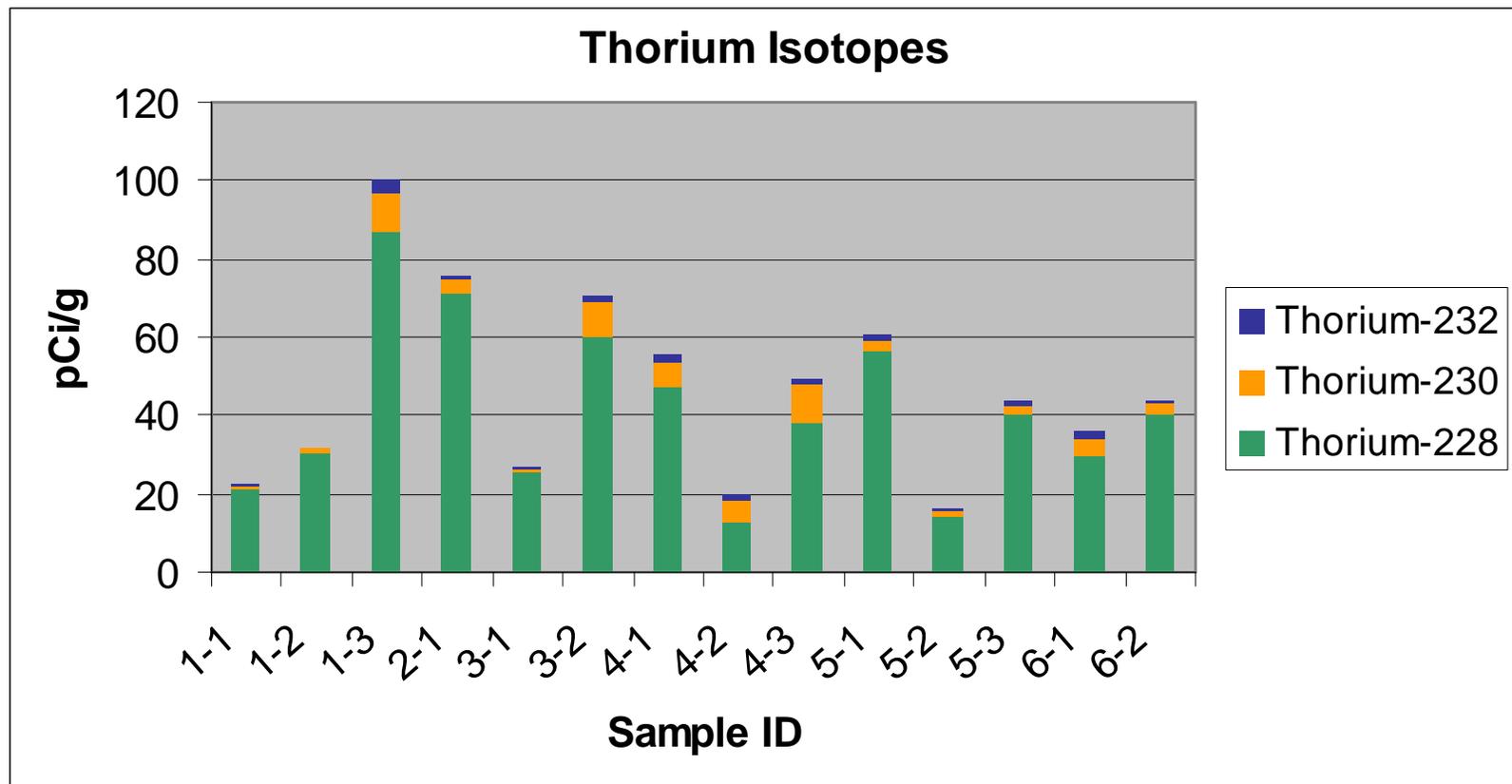
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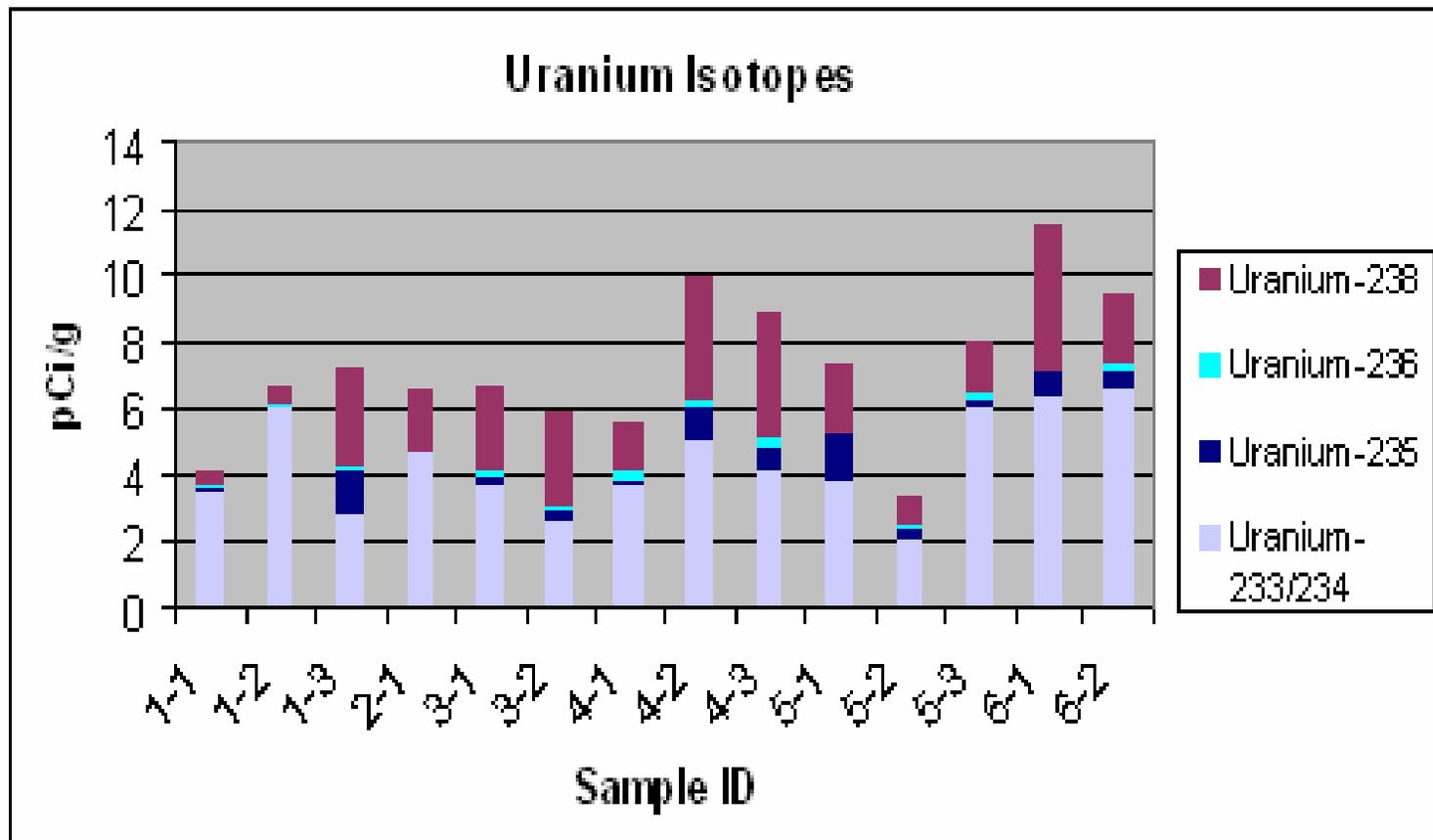
# Gross Alpha and Beta Radioactivity in Distribution System Solids



# Thorium in Distribution System Solids



# Uranium in Distribution System Solids



# Conclusions

- Arsenic and some radionuclides can concentrate in DW DS if in water at levels even below respective MCL
- The factors that determine how much accumulates are complex
- Also complex are the factors that impact release back into the distribution system. Disturbances to DS may release arsenic as well as factors such as:
  - *Particle destabilization*
  - *Desorption*
  - *Competitive desorption*
  - *Redox chemistry changes*
  - *Microorganisms*
- Health effects??
- Need for future investigation



# Thank-you

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